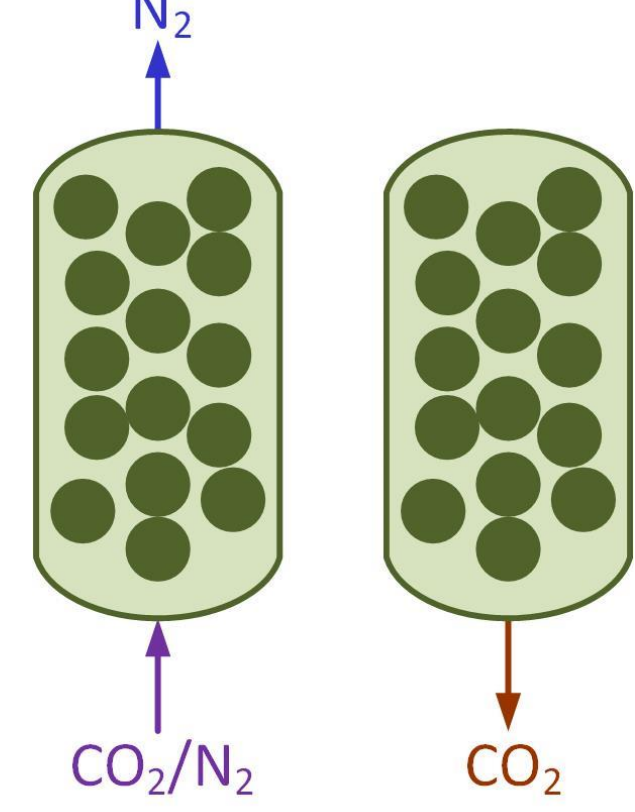


# Multivariate Data Analysis of Post Combustion CO<sub>2</sub> Capture

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## BACKGROUND

### Adsorption for CO<sub>2</sub> Capture



- Explosive growth in ability to synthesize novel solid sorbents
- Proven ability to reduce parasitic energy consumption
- Process design and optimization challenging

#### PSA Cycle Simulator

- Stiff PDEs
- Cyclic Process
- Convergence Complexities

**≥3000 single core hours for the evaluation of 1 material**

#### Optimizer

- Multi Objective Optimization
- Stochastic Output

## AIMS AND OBJECTIVES

**≥120 hours** to evaluate one material on multiple core processors



Previous Data about material performance

**3-4 orders of magnitude computing time saved Filtering out materials**



**≥0.12 hours** to evaluate one material on a single processor

## RESULTS

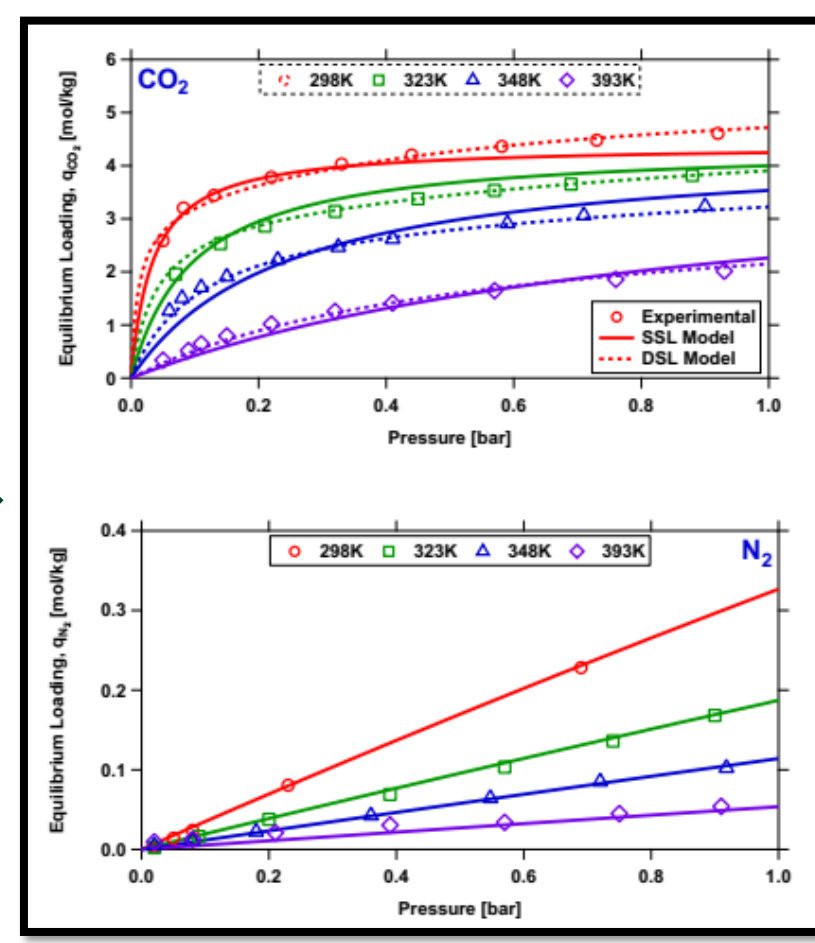
### Machine Learning Algorithm Training

#### Solid Adsorbent



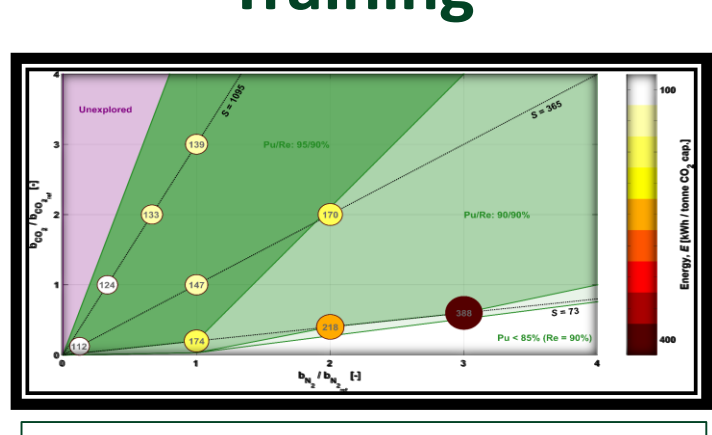
12 variables

#### Isotherm



4 variables

#### Training

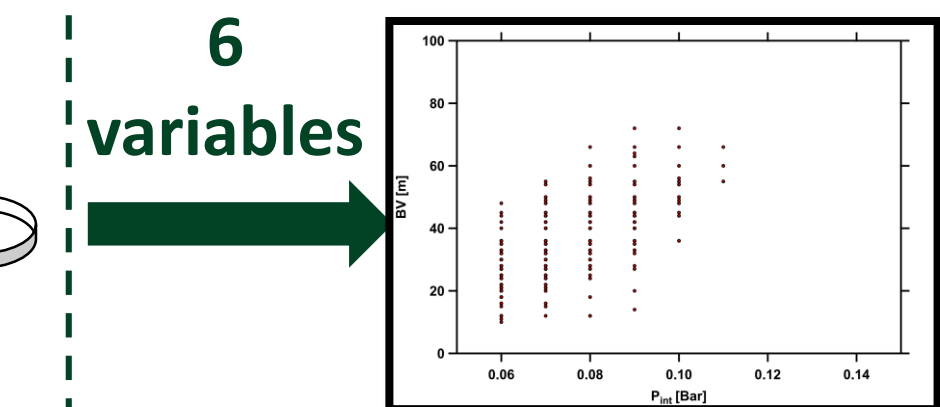


Simulation based Data

Supervised Machine Learning  
• Support Vector Machines  
• Ensemble Bagged

**R<sup>2</sup> Adj - 0.87  
Purity - 0.92  
Energy - 0.98  
Productivity - 0.99**

#### Dimensional Reduction (PCA)



6 variables

3 variables

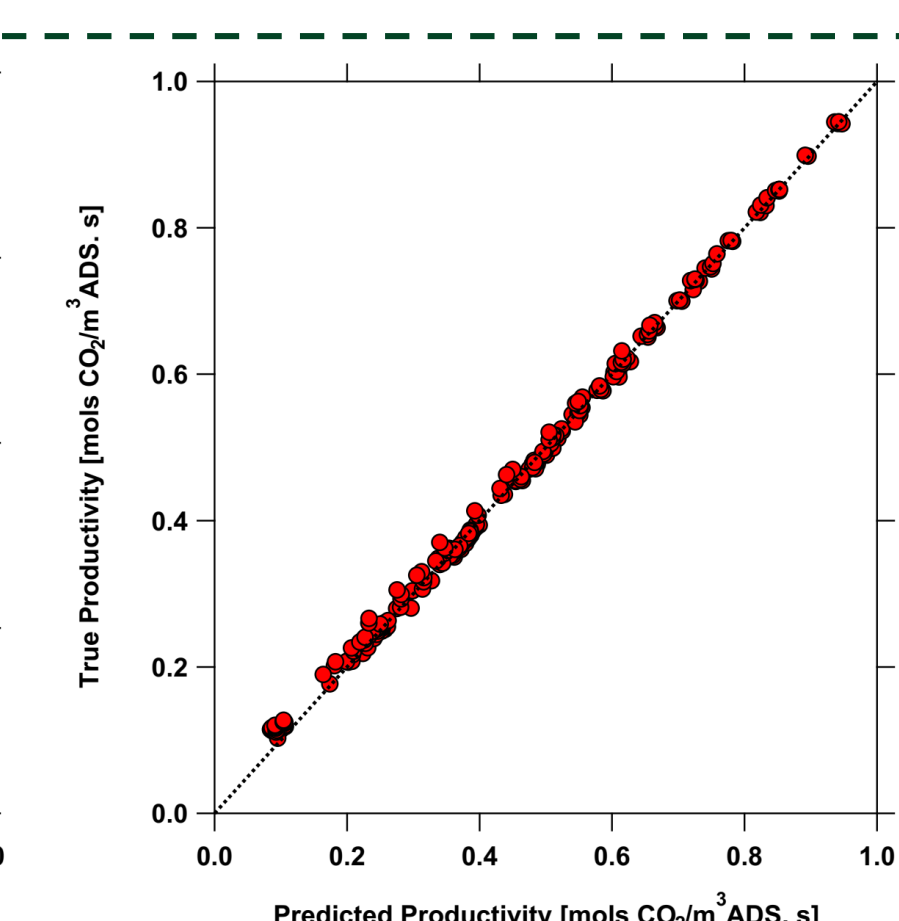
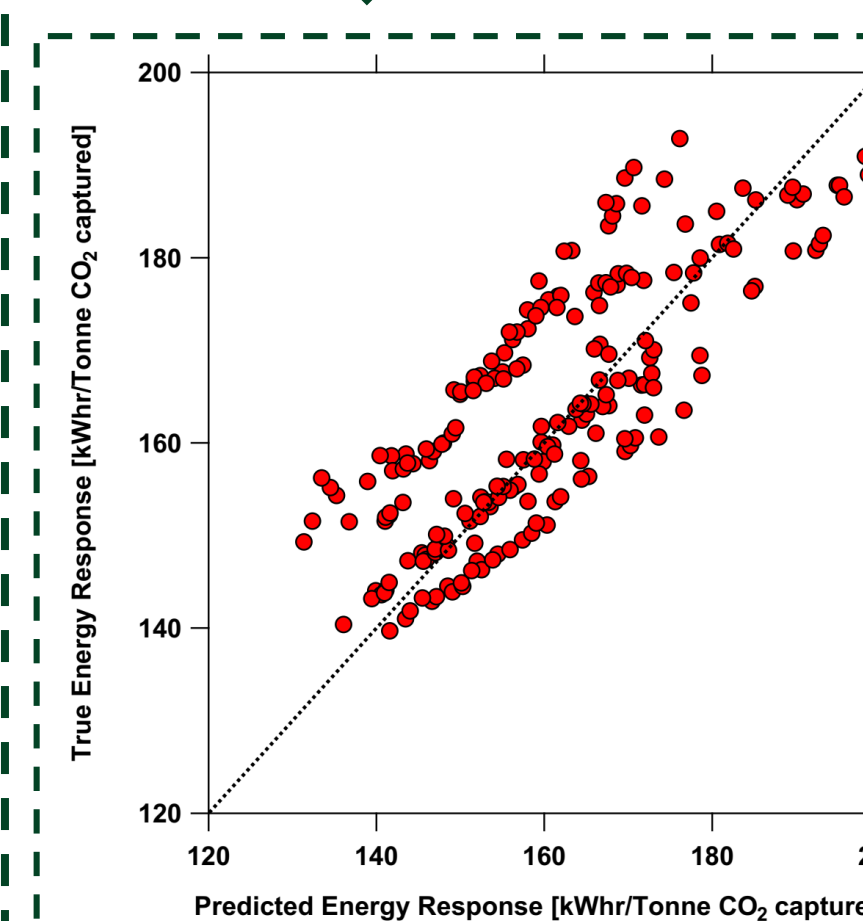
70 Materials for Evaluation

7 Variables

Proxy Model

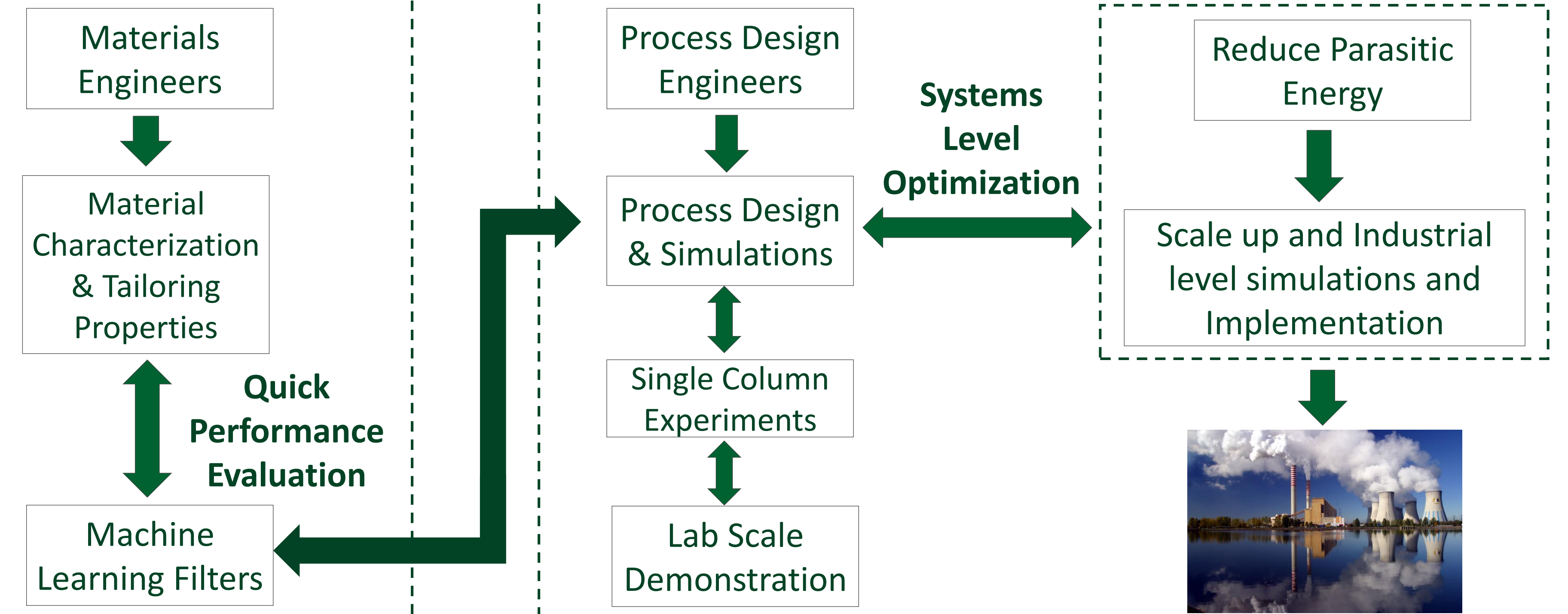
**Full Model Evaluation**  
150000 Core Hours  
• Accurate Purity, Recovery, Energy and Productivity values  
• 6 operational variables 12 material parameters

**Proxy Model Evaluation**  
1.5 Core Hours  
• Accurate Filtration of Purity and Recovery  
• Precise Productivity Predictions  
• Limitations- Energy Predictions

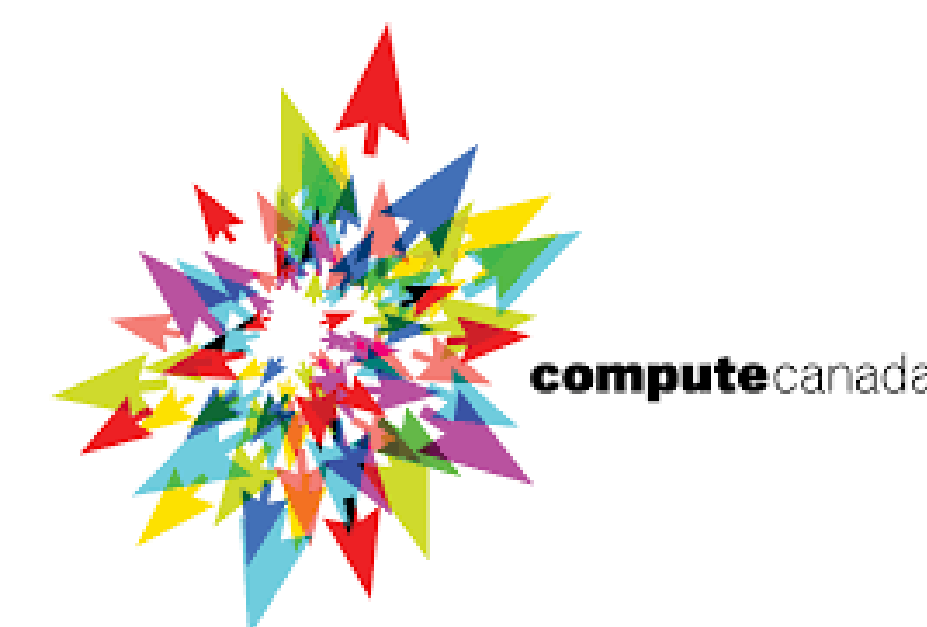


**Proxy Model Evaluation**  
**Pros:**  
• Quick and Accurate Evaluations of Multiple Materials  
• Quick Optimization of Materials  
• Based on only 4 Material properties  
**Cons:**  
• Not as Accurate as Detailed Model  
• Experimental Validation pending

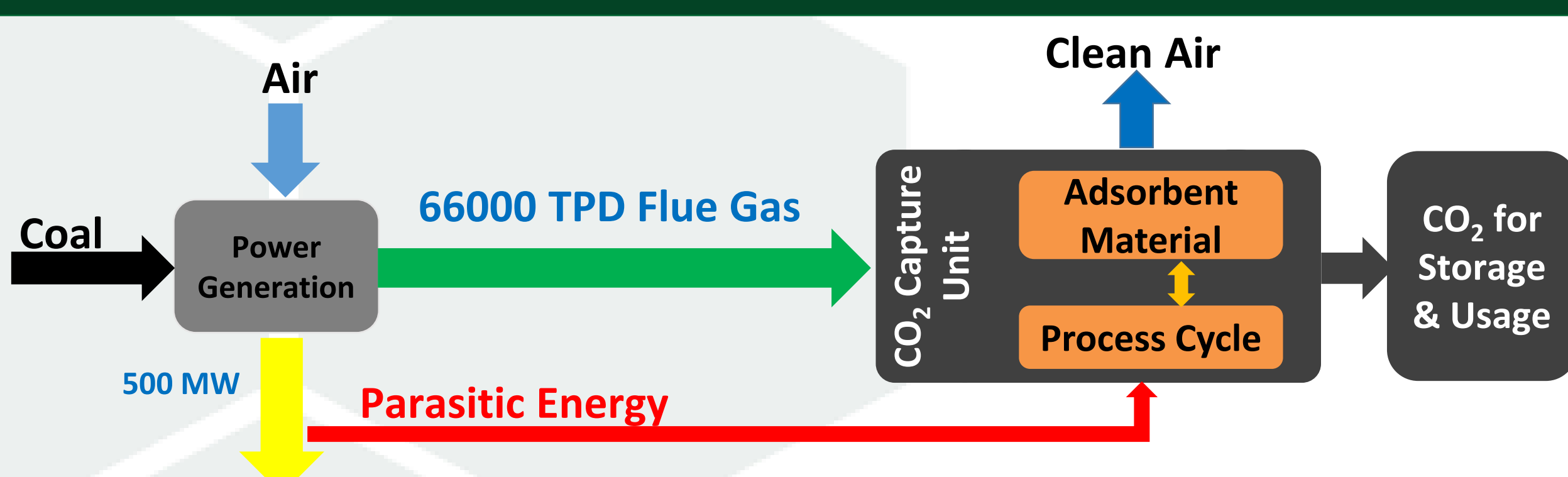
## FUTURE DIRECTIONS



## PARTNERS



## FES PROJECT OVERVIEW



Hydrocarbons will continue to serve as an essential energy source while the world transitions to a lower-carbon energy economy, but can we prevent the use of those fuels from contributing to the accumulation of CO<sub>2</sub> in the atmosphere? Existing technologies can capture carbon, but these methods can be costly and energy-intensive. Extracting energy without burning fuels, improving CO<sub>2</sub> capture efficiencies if they are burned, and finding effective ways to store or reuse captured carbon may be essential to ensuring it does not enter the atmosphere.